

Pergamon

Tetrahedron Letters, Vol. 35, No. 10, pp. 1531-1534, 1994 Elsevier Science Ltd Printed in Great Britain 0040-4039/94 \$6.00+0.00

0040-4039(94)E0099-J

X-RAY CRYSTAL STRUCTURE ANALYSES OF UNSATURATED BISSECO/BISHOMO DODECAHEDRANES

Manfred Keller, Klaus Scheumann, Klaus Weber, Torsten Voss, and Horst Prinzbach*

Chemisches Laboratorium der Universität Freiburg i. Br.

Institut für Organische Chemie und Biochemie, Albertstr. 21, D-79104 Freiburg

Hermann Imgartinger* and Uwe Reifenstahl

Organisch-Chemisches Institut der Universität Heidelberg, Im Neuenheimer Feld 270, D-69120 Heidelberg

Summary. For the sym-periplanar C,C double bonds in the bisseco dodecahedradiene 1 and the nonpentagonal dodecahedradiene 4 unusually short π,π -distances of 2.764 and 2.81 Å, resp., are determined (olefinic pyramidalization angles $\Phi = 14.6-18.3^{\circ}$). In the epoxy (dihydro) derivatives 2 (3), the respective values change to 2.83 (3.07) Å ($\Phi = 9.6-11.0^{\circ}$).

The bisseco dodecahedradienes **B**, arising from the pagodane precursors **A** through scission of the lateral (b) cyclobutane bonds,¹ are instrumental intermediates on the way to dodecahedranes **D** (route **B**; $\mathbf{B} \rightarrow \mathbf{C} \rightarrow \mathbf{D}$;² $\mathbf{B} \rightarrow \mathbf{E} \rightarrow \mathbf{F} \rightarrow \mathbf{D}^3$). A characteristic structural feature of these caged dienes, in comparison with the (seco)dienes **E** and **F**, is the unusually proximate syn-periplanar positioning of the two C_{*}C double bonds. The nature of the transannular bonding in the respective radical cations⁴ and dications,⁵ particularly the dependence of the σ -homoaromaticity observed for the bisseco dications \mathbf{B}^{2+} on geometry in going to the increasingly ball-shaped dications \mathbf{E}^{2+} and \mathbf{F}^{2+} , are prominent subjects of experimental and theoretical studies.⁶,⁷ In this context, precise



knowledge of structural situations, particularly of transannular distances, is of utmost importance. Specifically for the bisseco dienes **B** with their strong π,π -interaction, MM2 calculations⁸ - otherwise well reproducing the experimental structures of pagodanes⁹, dodecahedranes¹⁰ and secodienes¹¹ - had to be expected to underestimate the transannular π,π -distances (d₃).¹² X-ray structural analyses presented in this letter for bisseco 13,18-diketo-



3,8-diester-diene 1, its epoxy (2) and dihydro (3) derivatives 1^3 as well as for the non-pentagonal dodecahedradiene 4^{14} provide essential data, 1^5 even though in three cases crystal disorder had to be faced. The substitution patterns were generally established by the synthetic route.

Bissecodiene 1 crystallized from CH₂Cl₂ in small prisms¹⁶ and was found to be centrosymmetrically disordered with the 3,8-diketo-13,18-diester substitution pattern being superimposed. Hence, the C-C bond lengths and bond angles are average values, the thermal displacement parameters of the substituents are relative large. Still, the most important data, the distance (d₃) between the perfectly *sym*-periplanar π -bonds (interorbital angle 180°) and the transcaveal (d₁(2)) distances are unequivocal. The d₃ distance, with 2.764 Å one of the shortest of this type experimentally corroborated,¹⁷ is significantly longer than calculated (MM2) for the parent hydrocarbon **B** (2.62 Å, X = Y = CH₂), but only slightly so than calculated for the 3,8-diketone **B** (2.70 Å, X = C=O, Y = CH₂). In good agreement with the calculations are also the pyramidalization angles of the olefinic carbons (13.4° for the diketone). The length of the ridge-bonds (C5-C6) and the range of bond angles around the olefinic carbon atoms are expressions of the high molecular strain.¹²



Figure 1. SCHAKAL¹⁸ plots and selected X-ray structural data (bond lengths, through space distances (Å), bond, torsional, and pyramidalization (Ψ, Φ) angles (°)) for diene 1 (centrosymmetrical disorder).

Epoxyene 2 was crystallized from CH₂Cl₂/ethyl acetate¹⁶ and was found to be disordered about a molecular twofold axis perpendicular to the double bond C10-C11. Compared to 1, the transannular (d₃) distance is slightly longer, the transcaveal distances (d₁₍₂₎) are practically identical, the ridge C-C bonds (1.590 Å) again by far the longest. The pyramidalization angles at C10(11) are even (somewhat) smaller than in 1, at C1(20) closer to those of 1 than of 3, the epoxy side of the molecule is still rather flat.



Figure 2. SCHAKAL plots and selected X-ray structural data (bond lengths, through space distances (Å), bond, torsional, and pyramidalization (Ψ , Φ) angles (°)) for epoxyene 2 (twofold disorder).

Monoene 3 crystallized from $CH_2Cl_2/ethyl$ acetate in form of colorless prisms.¹⁶ As a consequence of the large pyramidalization angles at C10(11) - yet still much smaller than for saturated dodecahedrane (C₂₀H₂₀, 57.8°) - the d₃ distance is significantly enlarged; the transcaveal distances, however, are only insignificantly changed. The increase in strain tied to the hydrogenative saturation of one of the hyperstable C,C bonds in 1 is inter alia manifested in the average length (1.567 Å) of the bonds starting out from C10(11) and the large outer valence angles at the central carbons.



Figure 3. SCHAKAL plots and selected X-ray structural data (bond lengths, through space distances (Å), bond, torsional, and pyramidalization (Ψ , Φ) angles (°)) for monoene 3.

9,21-Dioxa-10,22-diketo-diene 4 crystallized from CH₂Cl₂/ethyl acetate as colorlesss needles¹⁶ and again turned out as centrosymmetrically disordered with the 10,22-dioxa-9,21-diketo form being superimposed. It is in line with calculations that the insertion of the two OCO-linkages into the skeletal periphery of the parent dodecahedradiene F ($d_{\pi,\pi} = 3.52$ Å; $\Phi = 42.8^{\circ}$)^{2,6} enforces a molecular curvature similar to that of 1. In fact, at close correspondence with respect to bond lengths and angles, the olefinic pyramidalization is somewhat stronger, the distance between the once again perfectly *syn*-periplanar π -bonds therefore somewhat larger than in 1.



Figure 4. SCHAKAL plots and selected X-ray structural data (bond lengths, through space distances (Å), bond, torsional, and pyramidalization (Ψ , Φ) angles (°)) for bislactodiene 4 (centrosymmetrical disorder).

Acknowledgment. Financial support by the Deutsche Forschungsgemeinschaft, the Fonds der Chemischen Industrie, and the BASF AG is gratefully acknowledged.

References and Notes

- 1) W.-D. Fessner, H. Prinzbach, The Pagodane Route to Dodecahedranes in Cage Hydrocarbons (Ed.: G.A. Olah), Wiley, New York, 1990, p. 353.
- 2) J.-P. Melder, R. Pinkos, H. Fritz, J. Worth, H. Prinzbach, J. Am. Chem. Soc. 1992, 114, 10213, and cit. lit.
- 3) R. Pinkos, J.-P. Melder, K. Weber, D. Hunkler, H. Prinzbach, J. Am. Chem. Soc. 1993, 115, 7173, and cit. lit.
- 4) H. Prinzbach, B.A.R.C. Murty, W.-D. Fessner, J. Mortensen, J. Heinze, G. Gescheidt, F. Gerson, Angew. Chem. Int. Ed. Engl. 1987, 26, 457.
- 5) G.K.S. Prakash, V.V. Krishnamurty, R. Herges, R. Bau, H. Yuan, G.A. Olah, W.-D. Fessner, H. Prinzbach, J. Am. Chem. Soc. 1988, 110, 7764.
- 6) K. Weber, H. Prinzbach, R. Schmidlin, F. Gerson, G. Gescheidt, Angew. Chem. Int. Ed. Engl. 1993, 32, 875.
- R. Herges, T. Bally, H. Prinzbach, unpublished results
 N.L. Allinger, J. Am. Chem. Soc. 1977, 99, 8127; N.L. Allinger, H.L. Flanagan, J. Comput. Chem. 1983, 4, 399; cf. W.v.E. Doering, W.R. Roth, R. Breuckmann, H.J. Figge, L. Figge, H.W. Lennartz, W.-D. Fessner, H. Prinzbach, Chem. Ber. 1988, 121, 1.
- 9) W.-D. Fessner, G. Rihs, H. Prinzbach, Tetrahedron Lett. 1983, 24, 5557; R. Pinkos, G. Rihs, H. Prinzbach, Angew. Chem. Int. Ed. Engl., 1989, 28, 303.
- 10) J.C. Gallucci, C.W. Doecke, L.A. Paquette, J. Am. Chem. Soc. 1986, 108, 1343; J.-P. Melder, H. Prinzbach, Tetrahedron Lett. 1990, 31, 1132.
- 11) H. Irngartinger, U. Reifenstahl, H. Prinzbach, R. Pinkos, K. Weber, Tetrahedron Lett. 1989, 30, 5459.
- 12) W.-D. Fessner, B.A.R.C. Murty, P.R. Spurr, R. Pinkos, J.-P. Melder, H. Fritz, H. Prinzbach, Chem. Ber. 1992, 125, 1697., B.A.R.C. Murty, R. Pinkos, P.R. Spurr, W.-D. Fessner, G. Lutz, H. Fritz, D. Hunkler, H. Prinzbach, *ibid.* 1992, *125*, 1719.
 13) J.-P. Melder, Dissertation, University of Freiburg, 1990.
 14) R. Pinkos, J.-P. Melder, H. Fritz, H. Prinzbach, *Angew. Chem. Int. Ed. Engl.* 1989, 28, 310.

- 15) For a detailed evaluation of theoretical and experimetal structural data s. F. Wahl, G. Lutz, W.-D. Fessner, H.-P. Melder, K. Weber, K. Scheumann, T. Voss, Chem. Ber., in preparation.
- 16) Crystal data for 1: Dimethyl 13,18-dioxo-nonacyclo[12.6.0.0²,6.0⁴,11.0⁵,9.0⁷,20.0¹⁰,17.0¹²,16.0¹⁵,19] icosa-1(20), 10-diene-3, 8-dicarboxylate, formula C24H20O6; crystal size 0.50 x 0.25 x 0.50 mm; space group P1 (No. 2), Z = 1, a = 8.333(1), b = 8.850(1), c = 6.204(1) Å; $\alpha = 98.77(1)^{\circ}$, $\beta = 106.31(1)^{\circ}$, $\gamma = 85.15(1)^{\circ}$, V = 433.6(2) Å³; D_{calc.} = 1.55 Mg m⁻³; Mo K_{α}, sin θ/λ = 0.61; unique reflections 2078, observed (I \ge 2.5 σ (I)) 1734; $\mathbf{R} = 0.067$, $\mathbf{R}_{\mathbf{W}} = 0.125$; parameters = 196.

Crystal data for 2: Dimethyl 13,18-dioxo-21- oxadecacyclo[12.7.0.01,20.02,6.04,11,05,9.07,20.010,17. 012,16,015,19]henicosa-10-ene-3,8-dicarboxylate, formula C24H22O6; crystal size 0.45 x 0.24 x 0.18 mm; space group P2₁/c (No. 14), Z = 4, a = 9.365(2), b = 11.563(1), c = 16.714(3) Å; β = 95.24(1)°, V = 1802.3(6) Å³; $D_{calc.} = 1.490 \text{ Mg m}^{-3}$; Cu K_{α}; sin $\theta/\lambda = 0.62$; unique reflections 3806, observed (I $\ge 2 \sigma$ (I)) 3149; R = 0.077, $R_W = 0.084$; parameters = 327.

Crystal data for 3: Dimethyl 13, 18-dioxo-nonacyclo[12.6.0.02,6.04,11,05,9.07,20.010,17.012,16.015,19] icosa-1(20)ene-3,8-dicarboxylate, formula C24H20O7, crystal size 0.4 x 0.2 x 0.1 mm; space group P21/c (No. 14), Z = 4, a = 8.791(2), b = 8.878(1), c = 23.27(3) Å; $\beta = 96.64(1)^\circ$, V = 1804.3(1) Å³; $D_{calc} = 1.548$ Mg·m⁻³; Cu K_{α}; sin $\theta/\lambda = 0.6226$; unique reflections 3800, observed (I $\ge 2 \sigma$ (I)) 2822; R = 0.045, R_w = 0.042, parameters = 359.

Crystal data for 4: 9,21-Dioxaundecacyclo[11.11.0.02,11.03,7.04,24.05,20.06,18.08,17.012,16.014,23] 015,19]tetracosa-3,15-diene-10,22-dione, formula C22H16O4; crystal size 0.35 x 0.14 x 0.60 mm; space group $P2_1/n$ (No. 14), Z = 2, a = 8.629(3), b = 10.553(2), c = 8.751(2) Å; B = 113.44(1)°, V = 731.1(2) Å³; $D_{calc.} = 1.56 \text{ Mg·m}^{-3}$; Cu K_a; sin $\theta/\lambda = 0.65$, unique reflections 1669, observed (I $\ge 2 \sigma(I)$) 1228; R = 0.087, $R_{w} = 0.065$; parameters = 147.

Further details of the X-ray structure analyses are available on request from the Fachinformationszentrum Karlsruhe, Gesellschaft für wissenschaftlich technische Information mbH, D-76344 Eggenstein-Leopoldshafen, on quoting the depository numbers CSD-400513 (1), CSD-400494 (2), CSD-400493 (3), and CSD-

- 400492 (4), the name of the authors, and the journal citation.
 17) K.B. Wiberg, M.G. Maturro, P.J. Okarma, M.E. Jeason, J. Am. Chem. Soc. 1984, 106, 2194; R.L. Viavattene, F.D. Greene, L.D. Cheung, R. Majeste, L.M. Trefonas, *ibid.* 1974, 96, 4342.
- 18) E. Keller, Schakal 92, University of Freiburg, 1992.

(Received in Germany 23 December 1993; accepted 7 January 1994)